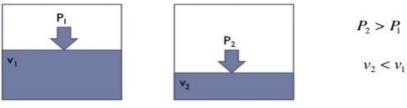
## Unit – I

# **Pressure and Pressure Measurement**

## Session – III

# Compressibility

- All fluids may be compressed by the application of external force, and when the external force is removed the compressed volumes of fluids expand to their original volumes.
- Compressibility is defined as it is the measure of change in volume when the substance is subjected to external pressure.



Change in volume (Decrease in volume)

- $dv = v_1 v_2$
- Also defined as the ratio of volumetric strain to compressive stress (increase in pressure).
- Compressibility of a fluid is quantitatively expressed as inverse of the bulk modulus of elasticity K of the fluid, which is defined as:

$$K = \frac{\text{Compressive stress (Increase in Pressure)}}{\text{Volumetric Strain}}$$

$$\mathbf{K} = \frac{\mathrm{d}\mathbf{p}}{-\left(\frac{\mathrm{d}\mathbf{v}}{\mathbf{v}}\right)}$$

- Since a increase in pressure always causes a decrease in volume, dv is always negative, and the minus sign is included in the equation to give a positive value of K.
- The SI unit of Bulk modulus of elasticity is N/m<sup>2</sup> or Pa.

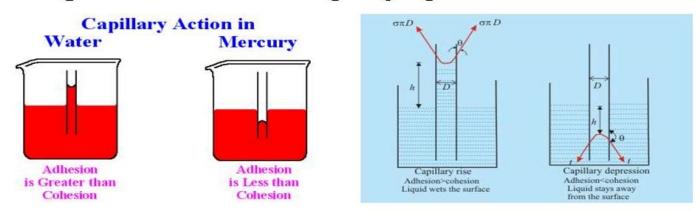
## Surface Tension and Capillarity

## A. Surface Tension

- Surface tension is due to cohesion between liquid particles at the surface
- Cohesion means inter-molecular attraction between molecules of the same liquid. That means it is a tendency of the liquid to remain as one assemblage of particles.
- The property of cohesion enables a liquid to resist tensile stress.



- The property of the liquid surface film to exert a tension is called the surface tension.
- Surface tension is also defined as it is the force required to maintain the unit length in equilibrium.
- It is denoted by σ (Greek 'sigma').
- In SI units surface tension is expressed in N/m.
- **B.** Capillarity
- Capillarity is defined as a phenomenon of rise or fall of a liquid surface in a small tube relative to the adjacent general level of liquid when the tube is held vertically in the liquid.
- The rise of liquid surface is known as capillary rise, while the fall of the liquid surface is known as capillary depression.



- It is expressed in terms of 'cm' or 'mm' of liquid.
- Its value depends upon the specific weight of the liquid, diameter of the tube and surface tension of the liquid.

$$h = \frac{4 \sigma \cos\theta}{\rho g d}$$

Adhesion and Cohesion

The tendency of two or more different molecules to bond with each other is known as **Adhesion**, whereas the force of attraction between the same molecules is known as **Cohesion**.

## **Problems**:

1. If a certain liquid has viscosity  $4.9 \times 10^{-3}$  N-sec/m<sup>2</sup> and kinematic viscosity  $3.49 \times 10^{-2}$  stokes, what is its specific gravity ?

Sol<sup>n</sup> :

Dy. Viscosity,  $\mu_l = 4.9 \times 10^{-3} \text{ N-sec/m}^2$ 

Kinematic Viscosity,  $v_1 = 3.49 \times 10^{-2}$  strokes = 3.49 x  $10^{-2} \times 10^{-4}$  m<sup>2</sup>/sec

Kinematic Viscosity,  $v_1 = 3.49 \times 10^{-6} \text{ m}^2/\text{sec}$ 

We know, Kinematic Viscosity =  $\frac{Dy. Viscosity}{Mass density}$ 

 $\begin{aligned} \textbf{i.e.} \quad \nu_l &= \frac{\mu_l}{\rho_l} \\ \textbf{There fore,} \quad \rho_l &= \frac{\mu_l}{\nu_l} = \frac{4.9 \text{ x } 10^{-3}}{3.49 \text{ x } 10^{-6}} = 1404.01 \text{ Kg/m}^3 \end{aligned}$ 

We know,Specific gravity of liquid = $\frac{Mass density of liquid}{Mass density of water}$ 

i.e.  $s_l = \frac{\rho_l}{\rho_w} = \frac{1404.01}{1000} = 1.404$ 

# 2. The volume of liquid is 2.5 m<sup>3</sup>. It is reduced by 0.0025 m<sup>3</sup> by increasing the pressure from 10 kpa to 210 kpa. Estimate the bulk modulus of elasticity of the liquid.

#### Soln :

Original Volume,  $V = 2.5 \text{ m}^3$ 

Change in Volume,  $-dv = 0.0025 \text{m}^3$  (-sign indicates reduction in volume)

Volumetric Strain,  $-\frac{dv}{(v)} = \left(\frac{0.0025}{2.5}\right)$ Increase in Pressure, dp = 210 - 10 = 200 kpa = 200 x 10<sup>3</sup> pa dp = 200 x 10<sup>3</sup> N/m<sup>2</sup> Bulk Modulus, K =  $\frac{dp}{-\left(\frac{dv}{v}\right)} = \frac{200 x 10^3}{\left(\frac{0.0025}{2.5}\right)}$ 

 $K = 2 \times 10^8 \, \text{N/m}$ 

3. The pressure of a liquid is increased from 500 Pa to 1300 Pa producing a decrease in volume of 0.16 percent. Find the bulk modulus of elasticity of the liquid.

Sol<sup>n</sup>: Original Volume, V = 100 percent Decrease in Volume, -dv = 0.16 percent *Volumetric strain*,  $-\frac{dv}{v} = \frac{0.16}{100}$ Increase in Preassure, dp = 1300 - 500 = 800 Pa = 800 N/m<sup>2</sup> *Bulk Modulus*,  $K = \frac{dp}{-\left(\frac{dv}{v}\right)} = \frac{800}{\left(\frac{0.16}{100}\right)} = 5 \times 10^5 N/m^2$